

Abstract

The origin of young Chang'e 5 (CE5) lunar basalts is highly debated. We present results from high-pressure, high-temperature (P - T) phase equilibria experiments, and from petrological modeling, to constrain the depth and temperature of the source of these unique mare basalts. The experimental results indicate that the CE5 basalts could have formed either by melting clinopyroxene and Fe-Ti oxide-rich cumulates in the shallow lunar mantle, or by extreme fractional crystallization of a hot Mg-rich parental melt. Our findings corroborate the local preservation of significant heat (of at least 1200 °C) in the lunar mantle that is needed to generate basaltic melts of CE5 compositions at 2 Ga. We argue that the CE5 basalts are most likely formed by melting of Fe and Ti-rich cumulates in the shallow lunar mantle as extreme fractional crystallization of olivine and plagioclase from picritic parental melts requires too high temperatures in the lunar mantle (> 1500 °C) at ~2 Ga.

Keywords: lunar mantle, fractional crystallization, lunar basalts, sample recovery mission, experimental petrology

1 Introduction

Recently, the Chinese CE5 mission (Che et al., 2021; Qian et al., 2021) returned new basaltic lunar rocks from the Oceanus Procellarum on the Moon. Geochemical studies on these samples, which were sampled in the young mare basalt unit Em4 in the Mons Rümker region (Qian et al., 2018), confirm the young age of these rocks (1.96–2.01 Ga, Che et al., 2021) that had already been dated by crater counting chronology (Hiesinger et al., 2003). This makes the CE5 samples the youngest studied igneous rocks from the Moon, about 1 Ga younger than the basalts from the US Apollo and Soviet Luna missions, and also all lunar basaltic meteorites (Elardo et al., 2014; Shearer et al., 2006). Therefore, the CE5 samples provide an excellent opportunity to study the evolution of the Moon through time.

33 To explain young lunar volcanism in the Procellarum KREEP (**K-REE-Phosphorous**) Terrain (PKT)
34 on the Moon, several possible heat sources or melting mechanisms have been suggested, which include a lunar
35 source that is rich in the KREEP (last remaining liquid after crystallization of the lunar magma ocean also
36 enriched heat-producing elements like U and Th) or water (Borg et al., 2004; Laneuville et al., 2013; McCubbin
37 et al., 2010). However, the CE5 basalts do not show a typical geochemical signature of that component (Tian
38 et al., 2021), and the hypothesis that young lunar basalts originated from a KREEP-rich mantle source (e.g.,
39 Borg et al., 2004; Laneuville et al., 2013) appears highly unlikely. Moreover, the CE5 samples contain little
40 water and other volatiles (Hu et al., 2021; Liu et al., 2022). Ziethe et al. (2009) proposed an insulating layer of
41 megaregolith, which reduced heat loss from the lunar mantle and hence extended the duration of volcanic
42 activity. In contrast, a thin crust (<30 km; Wieczorek et al., 2013) in the PKT region could lead to adiabatic
43 decompression melting and enable volcanism in that area late in the lunar history (Tian et al., 2021).

44 In this study we investigate the origin of the CE5 basalts using a two-fold approach. First, we
45 conducted high-pressure, high-temperature experiments to ascertain if the CE5 basalts could be primary,
46 unfractionated melts that have formed by melting of a lunar mantle cumulate.

47 Multiple saturation point (MSP) experiments as conducted in this study have proven useful to unravel
48 the origin of primitive lunar basalt samples such as picritic pyroclastic glasses (Chen et al., 1982; Green et al.,
49 1971, 1975; Hughes et al., 1988). These primitive samples (e.g., very low titanium A14 and A17 volcanic
50 glasses; Chen et al., 1982; A17 orange glass; Green et al., 1975) are multiply saturated with olivine (Ol) and
51 orthopyroxene (Opx) on their liquidus at ~2.0–2.5 GPa and ~1400–1500 °C. These findings indicate that the
52 picritic glasses represent near-primary basaltic melts derived from a mantle source that consists of olivine and
53 orthopyroxene within the lunar interior at depths greater than 400 km.

54 Second, we modeled the fractional crystallization of different primitive parental melt compositions to
55 establish, if the CE5 basalts could have formed by extensive fractionation of magmatic mineral phases such as

56 olivine and plagioclase. The CE5 basalts cannot be primary melts from an olivine-rich lunar mantle, since low
57 MgO/MgO+FeO ($Mg\# = 0.27$) usually indicate extensive fractionation of olivine and/or pyroxenes (Borg et
58 al., 2004; Che et al., 2021; Papike and Vaniman, 1978). Indeed, this is the preliminary interpretation for the
59 origin of the CE5 samples presented recently (Che et al., 2021; Tian et al., 2021). To further our understanding
60 of the evolution of the lunar mantle, and the generation of the Chang'e 5 basalts, we present new experiments
61 and fractional crystallization modeling.

62 **2 Methods**

63 **2.1. Starting material compositions**

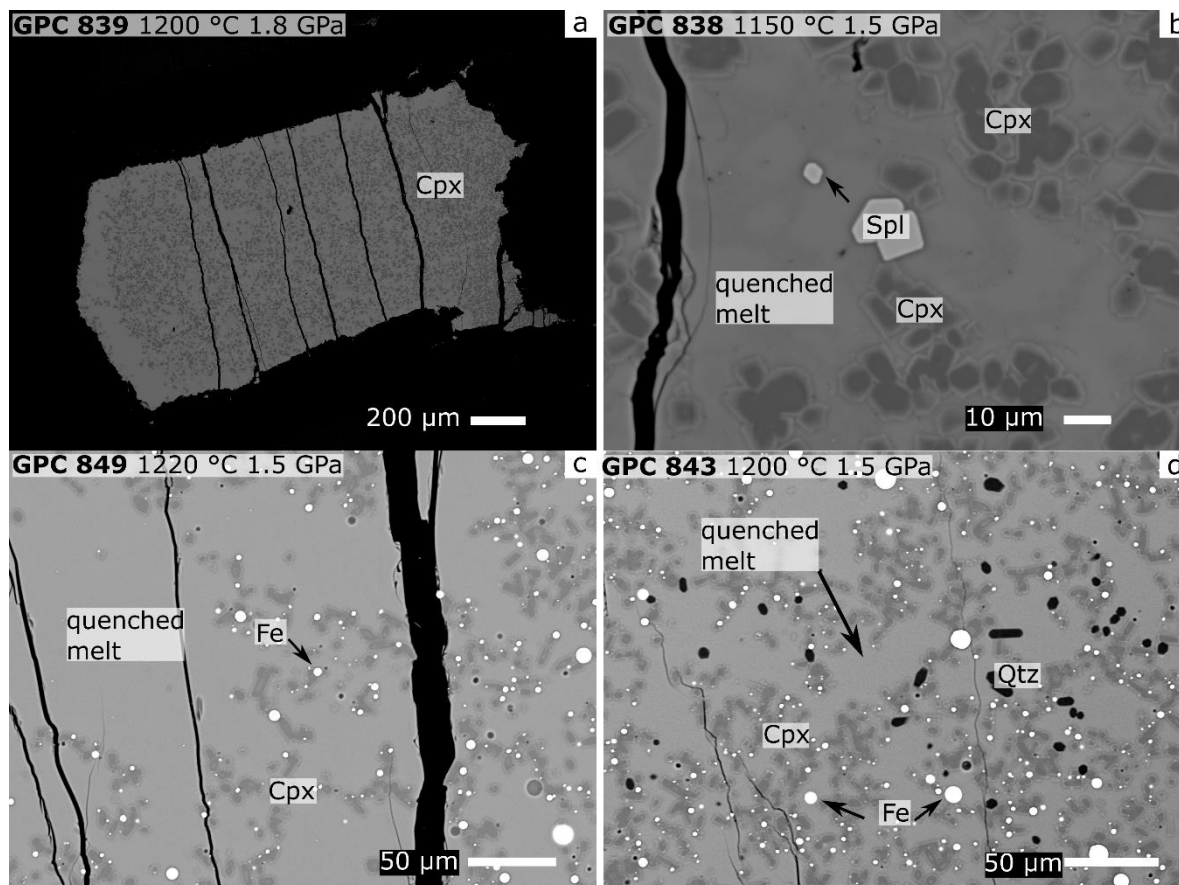
64 Two starting materials were prepared to match the bulk composition of CE5-B1 and CE5-B2 basalt
65 fragments (Che et al., 2021). These starting materials were synthesized by mixing reagent grade oxides and
66 carbonates (SiO_2 , Al_2O_3 , TiO_2 , MgO , Cr_2O_3 , $CaCO_3$, $MnCO_3$, K_2CO_3 , Na_2CO_3). Magnesium oxide was fired
67 at 1000 °C to remove hydroxide or carbonate, and the resulting MgO was always kept in a drying oven at
68 110 °C. The starting materials were homogenized in an agate mortar for 40 min under ethanol and the mixtures
69 were dried, placed in a Pt-crucible and decarbonated in a box furnace at 1000 °C for 10 h, followed by another
70 40 min of grinding and mixing. Iron was then added to the decarbonated mixture as hematite (Fe_2O_3) and P as
71 $H_9N_2O_4P$. The resulting Fe- and P-bearing mixtures were fused in a Pt-crucible at 1350 °C, reground and
72 reduced in graphite containers at 1100 °C in a vertical gas-mixing furnace (Gero GmbH, Germany) for 12 h,
73 using a steady flow of CO gas to impose an fO_2 corresponding to the graphite-CO-buffer and the final starting
74 material contained only Fe^{2+} .

75 The starting materials were always stored in a desiccator. To measure the starting material composition,
76 ~15 mg of the starting material was mounted on Re-wire loops and fused in a vertical gas-mixing furnace at
77 the iron-wüstite buffer (IW) and 1350 °C. The resulting glasses were embedded in epoxy resin, polished and
78 carbon coated for electron microprobe analysis. The chemical compositions of the starting materials together
79 with the lunar CE5 basalt compositions are shown in Table 1.

Table 1. Chemical compositions of CE5 basalts CE5-B1 and B2 and the experimental starting materials

Chemical composition (wt.%)	Chang'e 5 basalts		Experimental starting materials	
	CE5-B1	CE5-B2	BE-1 (n=10)	BE-2 (n=10)
SiO ₂	39.46–44.04	38.17–45.57	40.6(1)	43.0(2)
TiO ₂	4.37–7.91	5.84–9.68	5.7(2)	7.8(2)
Al ₂ O ₃	9.77–13.51	4.44–8.08	11.0(1)	7.03(7)
FeO	17.51–27.69	19.42–29.58	24.5(2)	22.7(1)
MnO	0.2–0.38	0.25–0.41	0.29(2)	0.38(2)
MgO	3.07–5.91	3.12–7.22	4.51(5)	5.45(7)
CaO	7.15–13.61	10.13–13.17	11.7(1)	12.23(9)
Na ₂ O	0.17–0.51	0.16–0.36	0.38(5)	0.35(5)
K ₂ O	0–0.19	0–0.18	0.11(2)	0.15(2)
Cr ₂ O ₃	0.03–0.25	0.05–0.19	0.37(5)	0.27(2)
P ₂ O ₅	0.06–0.36	0.08–0.26	0.19(4)	0.23(3)
Total			99.3(9)	99.5(8)

81 *The compositions experimental starting materials BE-1 and BE-2 determined using Electron Microprobe (EPMA)*
82 *with 1 σ standard deviations given in brackets and reported as last digits, e.g. 0.19(4) should read as 0.19 \pm 0.04 wt.%, n*
83 *= number of analyses*



84

85

86

Figure 1 Back-scattered electron (BSE) images of the run products from BE-1 (a–b) and BE-2 (c–d) starting materials compositions. (a) Run GPC 839 containing quenched melt and clinopyroxene (Cpx) at 1.8 GPa and 1200 °C. (b) Run GPC 838 contains quenched melt, clinopyroxene and traces of spinel (spl) (c) run GPC 849 contains quenched melt, clinopyroxene and Fe-metal blebs (d) Run GPC 843 contains small (<20 μm) euhedral quartz grains (Qtz), clinopyroxene, Fe-blebs and quenched melt

87 **2.2 Multiple saturation point (MSP) experiments**

88 The experiments between 1 and 2.5 GPa were performed in an end-loaded piston-cylinder apparatus (Boyd
89 and England, 1960). The starting materials were loaded into graphite capsules that were placed into ½ inch

90 piston-cylinder assemblies that consist of inner cylinders made of crushable alumina (6 mm outer diameter),
91 surrounded by a straight graphite furnace, a Duran glass cylinder (Schott, GmbH) and an outer talk sleeve. The
92 assembly was calibrated using the quartz-coesite transition (Bose and Ganguly, 1995) and the $\text{MgCr}_2\text{O}_4 +$
93 $\text{SiO}_2 = \text{MgSiO}_3 + \text{Cr}_2\text{O}_3$ reaction (Klemme and O'Neill, 1997) and resulted in a friction correction of -13% .
94 Based on our pressure calibration, the quoted pressures are accurate within 0.07 GPa. Experimental run
95 temperatures were monitored and controlled using a W-Re-thermocouple (Type D) and an Eurotherm
96 controller (Schneider Electric, Germany). Experiments were quenched by shutting off the electrical power.
97 The experimental conditions and results are given in Table 2 and exemplary back-scattered electron images,
98 taken with a scanning electron microscope, of the run products are shown in Figure 1.

Table 2. Experimental results of high-pressure piston cylinder (PC) experiments on compositions BE-1 and BE-2.

Run No.	Starting material	T [°C]	P [GPa]	Duration [hrs]	Phases	Modal proportions	Sum of sq. residuals
GPC 834	BE-1	1100	1	17	Cpx, FeTiOx+Ilm, Pl	64:13:23	1.352
GPC 841	BE-1	1130	1	17	Cpx, FeTiOx, Pl, melt	22:2:8:68	0.055
GPC 838	BE-1	1150	1.5	14	Cpx, FeTiOx, melt	17:<1:82	0.19
GPC 831	BE-1	1180	1	8	FeTiOx, melt	<1:99	0.76
GPC 844	BE-1	1200	1.5	8.5	Cpx, melt	10:90	0.125
GPC 839	BE-1	1200	1.8	7	Cpx, melt	16:84	0.23
GPC 827	BE-1	1250	1	14	melt (+ Fe)	99:<1	0.136
GPC 837	BE-1	1300	2.5	6	melt	100	0.144
GPC 853	BE-1	1180	1.4	8.5	Cpx, melt	10:90	1.4
GPC 856	BE-1	1180	1.2	8.5	FeTiOx, melt	<1:99	0.497
GPC 862	BE-1	1160	1.3	9	FeTiOx, Cpx, melt	<1:17:82	1.09
GPC 863	BE-2	1180	1.2	8	Fe-metal, Cpx, Qtz, melt	12:18:1:69	1.2
GPC 842	BE-2	1150	1	17	Fe-metal, Qtz, Cpx,	10:6:37: 7:7:<1:33	0.03

					Ilm, Pl, Ttn, melt		
GPC 843	BE-2	1200	1.5	8	Fe-metal, Qtz, Cpx, melt	12:1:18:69	0.21
GPC 848	BE-2	1200	1	10	Fe-metal, Cpx, melt	11:4:85	0.39
GPC 849	BE-2	1220	1.5	7	Fe-metal, Cpx, melt	12:10:78	0.713
GPC 851	BE-2	1250	1	8.5	Fe-metal, melt	13:87	0.134

99 *Phase proportions determined by least squares linear regression of analyzed phases (Supplementary Material) per run*
100 *against starting composition. Phase abbreviations: clinopyroxene = Cpx, plagioclase = Pl, quartz = Qtz, titanite = Ttn, Fe-Ti-oxides*
101 *(FeTiOx) comprising ulvöspinel-chromite-spinel solid solutions, and ilmenite (Ilm), Fe= metallic iron, melt = quenched melt(see also*
102 *the Supplementary Material), T= temperature in °C, P = pressure in GPa. Modal proportions were calculated using the bulk*
103 *compositions and mineral compositions.*

104 All experimental run products were mounted in epoxy-resin and polished using different diamond
105 pastes. The mounts were then carbon coated, examined with a JEOL JSM 6510-LV scanning electron
106 microscope and analyzed quantitatively with an electron probe microanalyzer (EPMA; JEOL JXA 8530 F
107 Hyperprobe) at the Institute for Mineralogy, University of Münster (Germany).

108 **2.3 Electron Probe Microanalyses (EPMA)**

109 All EPMA measurements were performed using an acceleration voltage of 15 kV and a beam current of 15 nA
110 except for plagioclase (Pl), which was measured with a beam current of 10 nA. The beam diameters for glass
111 analyses were set according to the size of the melt pockets in the experiments which range from 5 to 20 μm .
112 Silicate and oxide minerals were measured with a beam diameter of 1–5 μm . The matrix corrections were
113 made according to the $\phi(\rho z)$ procedure (Armstrong, 1991). Details can be found in the Supplementary Material.

114 **2.4 Attainment of equilibrium and assessment of $f\text{O}_2$ in the experiments**

115 The durations of the experiments (7–17 h) was sufficient to produce compositionally homogeneous minerals
116 (Singletary and Grove, 2008; van Orman and Grove, 2000). Minerals in our runs are small and generally
117 unzoned; only clinopyroxene contains small quench rims (Fig. 1b–d). The standard deviations of the average
118 compositions of mineral and melts are very low, which is consistent with the attainment of chemical
119 equilibrium in these runs. A profile through a large (80 μm) clinopyroxene grain confirms homogeneity
120 (Supplementary Material Fig. 1).

121 The $f\text{O}_2$ for runs on BE-1 was determined in run GPC 831 by adding an Fe-Ir alloy $f\text{O}_2$ -sensor
122 (resulting in IW+2). We assume that the $f\text{O}_2$ is identical in all runs on that composition. All experiments with
123 BE-2 contain Fe-blebs, which indicate a lower $f\text{O}_2$ (Sato et al., 1973). The $f\text{O}_2$ for BE-2 was estimated using
124 the equilibrium of $2\text{Fe} + \text{O}_2 = 2\text{FeO}$ and resulted in IW–1.5 (for details refer to the supplementary material).
125 The $f\text{O}_2$ observed in the experiments is consistent with the current best estimates on $f\text{O}_2$ derived from lunar
126 basalt samples (Sato et al., 1973; Wadhwa, 2008). Since run conditions and experimental methods are very
127 similar, we attribute the difference in $f\text{O}_2$ between experiments conducted on BE-1 and BE-2 to the different
128 chemical compositions of the starting materials BE-1 and BE-2.

129 **3 Results**

130 **3. MSP experiments**

131 We performed 17 PC-experiments on starting materials BE-1 and BE-2, representing the CE5 basalts CE5-B1
132 and CE5-B2 from Che et al. (2021), and exemplary BSE images of the runs are shown in Fig. 1. Table 2 gives
133 the experimental run conditions, stable mineral phases, and calculated phase proportions in each run. More
134 details are given in the Supplementary Material. The high-pressure phase relations for the two compositions
135 are shown in Fig. 2. For composition BE-1, clinopyroxene (Cpx) is the only liquidus phase between 1.4–
136 1.7 GPa and 1180–1200 °C (Fig. 1a; Fig. 2a), and at lower pressures clinopyroxene, plagioclase, Fe-Ti oxides
137 (ulvöspinel-chromite-hercynite-spinel), and ilmenite appear. Fe-Ti oxides are found in some high temperature,
138 low pressure runs (GPC 838, 831, 856; Table 2). As the position of the MSP can only be bracketed by our

139 experiments, we infer a MSP at pressures between 1.2 and 1.4 GPa, and at temperatures between 1200 and
 140 1250 °C, whereas melts are saturated in both, clinopyroxene and Fe-Ti oxides.

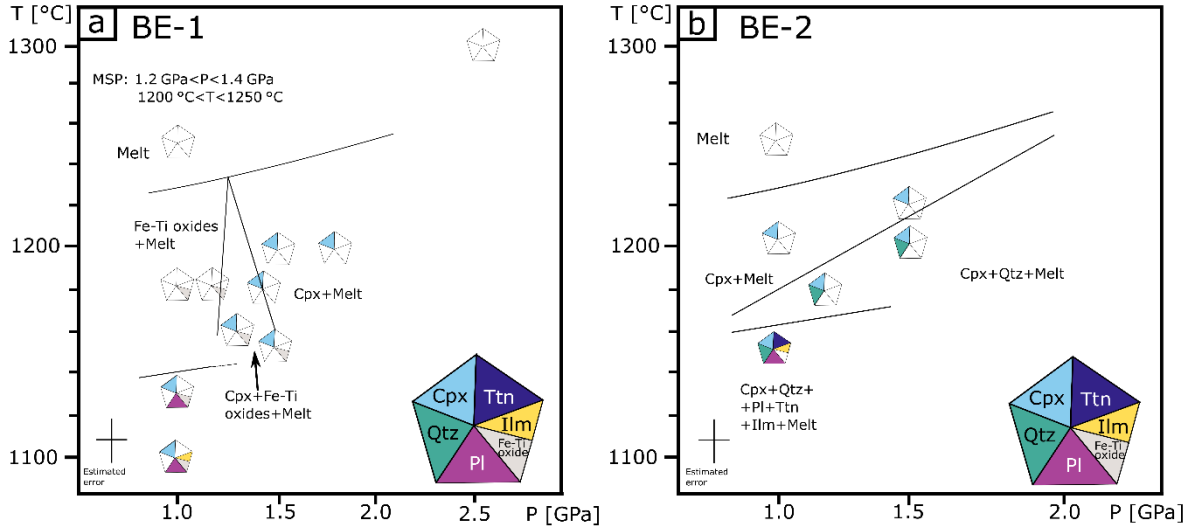


Figure 2 Pressure-temperature diagram showing the experimental results on the starting materials BE-1 (CE5-B1) (a) and BE-2 (CE5-B2) (b) between 1–2.5 GPa and 1100–1300 °C. Stable mineral phases are depicted in pentagons at the individual P-T conditions: clinopyroxene (Cpx), Fe-Ti oxides (solid solutions of the ulvöspinel-chromite-hercynite-spinel series), quartz (Qtz), plagioclase (Pl), titanite (Ttn), and ilmenite (Ilm). (a) BE-1 is multiple saturated with clinopyroxene, and Fe-Ti oxides at about 1.3 ± 0.1 GPa and 1225 ± 25 °C. (b) BE-2 has only Cpx on the liquidus between 1200–1250 °C and 1–1.5 GPa, but 40–50 °C below the liquidus, Qtz appears at 1.5 GPa and followed by Pl, Ilm, and Ttn at 1 GPa and $T = 1180$ °C. Note the estimated uncertainties in P and T indicated in (a) and (b).

141

142 Experimental results for the BE-2 starting material are significantly different, as in these runs (Fig. 1c–d, Fig.
 143 2b, Table 2) only clinopyroxene occurs as a liquidus phase (liquidus at 1–1.5 GPa and 1200–1220 °C, Fig. 1c),
 144 followed by quartz at 1.5 GPa and 1200 °C (Fig. 1d). Clinopyroxene, quartz, plagioclase, and ilmenite
 145 crystallize at 1 GPa and 1150 °C. We conclude that no clear-cut multiply saturated phase assemblage exists
 146 for the BE-2 composition, but it should be noted that multiple phase saturation occurs only ca. 40 °C below
 147 the liquidus. Furthermore, some runs with BE-1 composition contain sub-micron sized Fe-rich metal grains,
 148 which are not significantly affecting mass balance. In contrast, the experiments on CE5-B2 composition
 149 contain considerable Fe-metal blebs in all of the experiments (Fig. 1c–d, Table 2; Supplementary Material).

150

151 4 Discussion

152 4.1. Can the Chang'e 5 basalts be primary melts from Fe- and Ti-rich cumulates?

153 Our experimental results on the synthetic analog of CE5-B1 show that the system is saturated on the
154 liquidus with clinopyroxene, and Fe-Ti oxides at a pressure of 1.3 ± 0.1 GPa and 1225 ± 25 °C with plagioclase
155 appearing shortly after (~ 1150 °C; Fig. 1a). The crystallization of either ilmenite or Ti-bearing spinel
156 (Ulvöspinel) as a Fe-Ti-bearing oxide depends on fO_2 (Schmitt and Kraettli, 2022). Since both are present in
157 our experiments, we infer a MSP of clinopyroxene and FeTi-oxides. In contrast to most previous studies on
158 lunar samples defining multiple phase saturation, neither orthopyroxene, nor olivine are present in our
159 experiments. The mineral assemblage at multiple phase saturation of the CE5-B1 basalt is also significantly
160 different from most other picritic lunar basalts, which contain olivine, spinel and low-Ca pyroxene at the MSP
161 (Elardo et al., 2015; Green et al., 1975; Grove and Vaniman, 1978). The multiple saturation points for primitive
162 (lunar and terrestrial) basalts must be regarded as minimum pressures and temperatures of origin (e.g., Longhi,
163 1992a). If the CE5 basalts are primary, unfractionated melts, the MSP indicates that the melt was in equilibrium
164 with clinopyroxene, and Fe-Ti oxides at ~ 250 km depth within the lunar mantle, following the pressure-depth
165 relationship of Elkins-Tanton et al. (2011). This is in stark contrast to the lunar picrites which originated from
166 harzburgite rocks much deeper in the lunar mantle, i.e. at $1.5\text{--}2.5$ GPa and $1400\text{--}1500$ °C (Chen et al., 1982;
167 Hughes et al., 1988; Wagner and Grove, 1997), which corresponds to a depth of $300\text{--}600$ km. A mantle
168 cumulate from which the CE5 basalts originated has to be of different character. Our experiments show that
169 the CE5-B1 basalt might be a primary melt from a late stage cumulate containing clinopyroxene and Fe-Ti
170 oxides. The pressure indicated by our experiments (~ 1.2 GPa) locates these cumulate at a depth of at least
171 250 km within the lunar mantle, which supports that the late stage cumulates did not persist at their original
172 depth of ~ 100 km (Snyder et al., 1992), but sank further into the mantle, as proposed by the overturn hypothesis
173 (e.g., Green et al., 1975; Spera, 1992; Ringwood and Kesson, 1976; van Orman and Grove, 2000; Wagner and
174 Grove, 1997).

175 Our MSP is further supported by experimental investigations on near-liquidus phase relations of high-
176 Ti, high-Fe cumulates yielding very similar phase assemblages with first clinopyroxene and clinopyroxene +
177 Fe-Ti oxides crystallizing between $1100\text{--}1200$ °C at $0\text{--}1.3$ GPa. Additionally, the coexisting melt

178 compositions are similarly enriched in FeO and depleted in MgO (van Orman and Grove, 2000). Even the
179 appearance of quartz is reported in these experiments, which we find in the experiments on the CE5-B2
180 experimental analogue. Experiments with the CE5-B2 composition (i.e., BE-2), however, do not show a MSP
181 (Fig. 2b). Nevertheless, the near-liquidus phases are similar to composition B1 and we conclude, that the
182 former composition is not far from being multiple saturated. According to the petrographic evaluation of both
183 natural CE5 samples (Che et al., 2021 Electronic Supplementary Material), B1 seems to be the less-altered
184 sample and, according to its major element composition, is generally more representative for the young Em4
185 unit, sampled during the ChangE5 mission (Li et al., 2022). Thus, we regard the MSP found for B1 composition
186 as most significant for the CE5 basalts.

187

188 **4.2. Are the Chang'e 5 basalts products of extensive fractional crystallization from picritic** 189 **parental melts?**

190 The experimental evidence for CE5 basalts being primary melts of a shallow source (~250 km) is only
191 one of the hypotheses discussed in recent studies (e.g., Zhong et al., 2022). Based on the high FeO content
192 (~23 wt.% FeO) and low Mg# (~0.27), several studies propose that these basalts are not primary melts, but the
193 product of extensive fractional crystallization (Che et al., 2021; Tian et al., 2021). For example, Tian et al.
194 (2021) suggested low-degree (2–3 %) partial melting from a harzburgite mantle followed by extensive (43–
195 78 %) fractional crystallization as a possible scenario.

196 To critically evaluate our experimental findings, we explore here if fractional crystallization can be a
197 way to produce the CE5 basalts, and if we can make any reasonable assumptions about the composition of the
198 potential parental magma of these basalts. Specifically, we tested whether the CE5 basalts can be derived from
199 lunar picritic melts (A14yellow, A17orange A15red: Delano, 1986; A17yellow: Delano and Lindsley, 1983;
200 A15yellow: Hughes et al., 1988). These samples, tiny glass beads in the lunar soils, are of volcanic origin and
201 likely pristine samples of the lunar interior (Delano, 1986) and hence, may be good candidates for parental
202 melts of the CE5 basalts. We chose the most primitive glasses, according to their MgO. These primitive

203 compositions were experimentally proven to be melts that originate from the deep lunar interior (Krawczynski
 204 and Grove, 2012; Brown and Grove, 2015). The description of the fractional crystallization model and the
 205 normalized major element compositions of these glasses are listed in the Electronic supplementary material.
 206

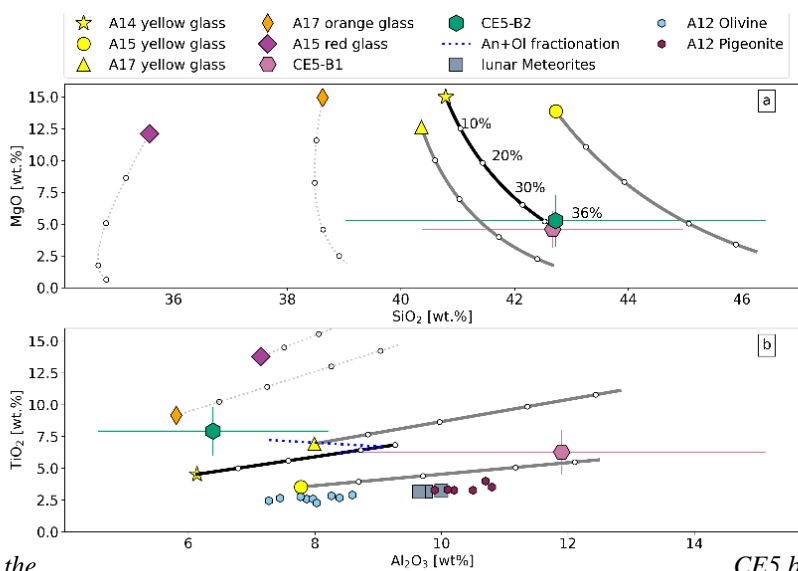


Figure 3. Variation diagrams showing the liquid line of descent of possible parental melt compositions upon fractional crystallization of equilibrium olivine (Ol) and anorthite (An, blue dotted line). Also given are the two CE5 basalt compositions CE5-B1 and CE5-B2 (hexagons) (a) MgO vs SiO₂: Ol fractionation only (grey lines) of primary Apollo glass melts. The results show that the A17 orange and A14 red glass compositions cannot be parental melts of

the CE5 basalts, but fractional crystallization of olivine from the yellow glass melt compositions lead to the CE5 basalts (within the uncertainties) (b) TiO₂ vs Al₂O₃: Ol only fractionation of a A14 yellow glass parental melt leads to a melt that is (just) within the uncertainties of the CE5-B1 composition, but additional fractionation of plagioclase (blue stippled line) is needed to move the Al₂O₃ of the fractionating melt composition towards the compositional range of the CE5-B2 sample. Additional information added are A12 Ol basalts (blue hexagons), A12 Pigeonite (Pig) basalts (purple hexagons), and lunar basaltic meteorites NWA4734, 032 and LAP02205 (all together referred to as lunar meteorites; grey squares; plotted compositions taken from Elardo et al. 2014 and Day et al. 2006 and references therein). Natural samples fall on the Ol fractionation trendline from yellow picritic glasses towards CE5 basalts. A15 yellow glass, lunar meteorites and A12 basalts appear to have fractionated from a similar source (Elardo et al., 2015)

207 Melt evolution trends are caused by olivine and minor plagioclase fractional crystallization (Fig. 3; Longhi,
 208 1992b; Rhodes et al., 1977). As shown in Fig. 3 (a), the fractional crystallization of olivine alone (grey solid
 209 lines) from primitive parental melts lead towards lower MgO in the evolved melts. However, the orange glass
 210 composition and the red glass composition are not suitable candidates for parental melts of the CE5 basalts
 211 (Fig. 3). Fractional crystallization of olivine from parental melt with yellow glass composition (A14, A15, and
 212 A17) yields a melt composition that is within the range of the CE5 samples (Fig. 3, Electronic Supplementary

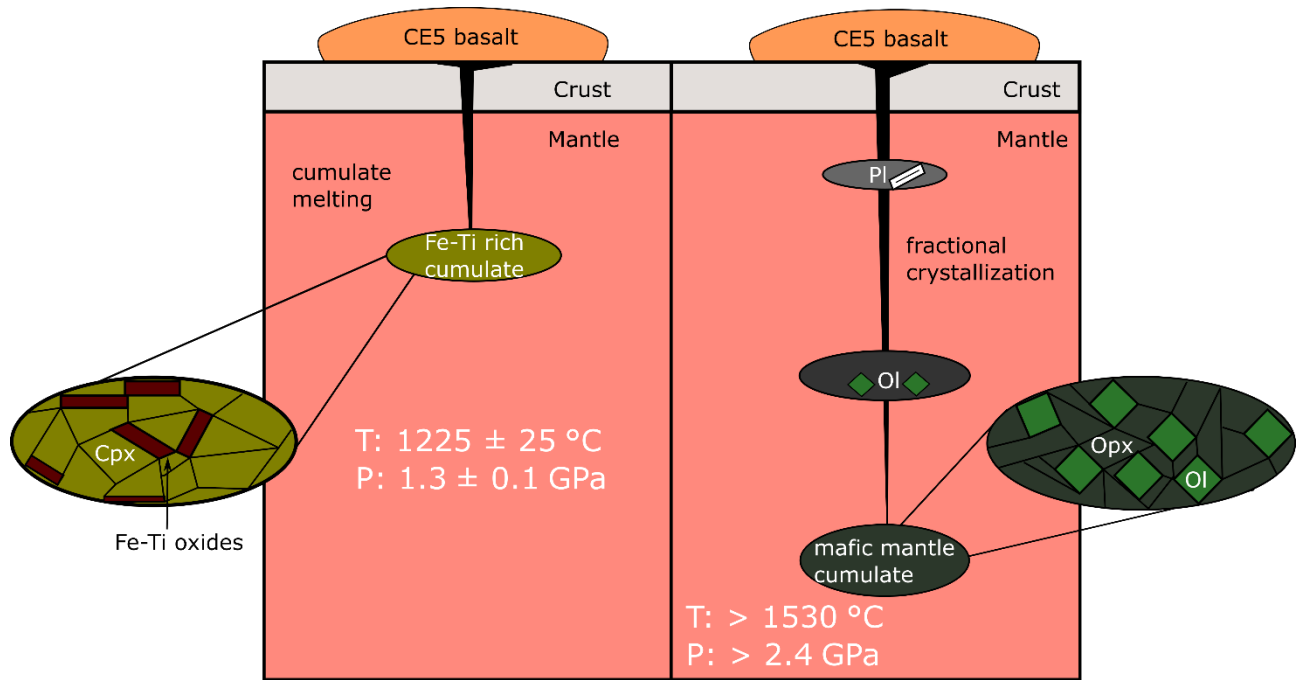
213 Fig. 2). The modeling outcome for both investigated CE5 compositions, however, is slightly different. The
214 fractional crystallization path in Figure 3 shows, that about $36\pm 8\%$ Ol crystallization from a parental melt
215 with primitive A14 yellow glass composition can lead to a melt composition within the range of the CE5-B1.
216 On the other hand, to move the fractionation line closer to the low-alumina composition CE5-B2, which is not
217 achieved in the $\text{TiO}_2\text{-Al}_2\text{O}_3$ space by the fractionation of olivine alone (Fig. 3b), additional fractionation of
218 plagioclase together with olivine (blue dotted line Fig. 3b) is required. Specifically, that includes initial 33 %
219 olivine fractionation followed by $8\pm 4\%$ olivine and plagioclase fractionation to shift the melt composition
220 towards the CE5-B2 (depicted as the blue stippled line in Fig. 3b). Our suggestion for additional plagioclase
221 fractionation is supported by the fact that plagioclase can appear after olivine fractionation towards ambient
222 pressures (Elardo et al., 2015; Grove and Vaniman, 1978) in low-Ti mare basalts. We argue, that CE5-B2
223 composition could have experienced a slightly faster near-surface cooling to initiate plagioclase crystallization.
224 The TiO_2 contents of both CE5 compositions lie within the liquid composition of a fractionated parental melt
225 of A14 yellow glass composition after $\sim 36\%$ olivine fractionation. (Fig. 3b). In order to produce Fe enriched,
226 Mg depleted young lunar basalts, olivine fractionation from a primitive parental mantle source seems to be a
227 viable process (e.g., Elardo et al., 2015). The CE5 basalts could be products of olivine fractionation from a
228 primitive source - a similar process potentially responsible for the genesis of the ~ 1 Ga older lunar meteorite
229 samples (LAP02205, NWA4734, NWA032; compiled major element compositions in Elardo et al., 2015) and
230 the A12 low-Ti basalts (Fig. 3b).

231

232

233 **4.3. On the thermal state of the lunar mantle at the time of CE5 basalt generation**

234



235

236 *Figure 4 Diagrams showing hypothesis (a) and (b) of the origin of young CE5 lunar basalts at 2 Ga. (a)*
 237 *According to the experimental results, the CE5 basalts are the result of melting of a late stage mantle cumulate (Fe-Ti*
 238 *rich cumulate) consisting of clinopyroxene (Cpx) and Fe-Ti oxides at T 1225±25 °C at a depth of ~250 km. (b) Fractional*
 239 *crystallization of olivine (Ol) and minor fractions of plagioclase (Pl) from a primitive melt of a harzburgite lunar mantle*
 240 *cumulate (P and T taken from MSP experiments of Brown and Grove, 2015). A14 yellow glass, the potential parental*
 241 *melt of CE5 basalts, is at multiple saturation with orthopyroxene (Opx) and olivine (Ol) at ~500 km depth and 1530 °C,*
 242 *the conditions where the melt would have started its fractionation path to the lunar surface, if the CE5 samples would be*
 243 *product of fractional crystallization.*

244

245 In the previous sections we have introduced two possible scenarios that can explain the origin of the young
 246 CE5 basalts.

247 First, experimental results provide constraints that the CE5 basalts could have originated as melts from a late
 248 stage mantle cumulate with a MSP at 1.3±0.1 GPa and 1225±25 °C (Fig. 4a). Second, the CE5 basalts can
 249 alternatively be explained as the product of extensive fractional crystallization from a primitive picritic melt
 250 with a composition similar to the A14 yellow glass (Fig. 4b). Here, we discuss which of these models is more
 251 favorable based on the current understanding we have about the thermal evolution of the lunar mantle.

252

253 If we consider the experimentally determined conditions of multiple phase saturation of the CE5
254 samples as the minimum depth and temperature of origin, where they were last in equilibrium with a mantle
255 melt, the young CE5 basalts would have originated from a depth of at least 250 km within the lunar mantle
256 and at temperatures of at least 1200 °C. This would be an extraordinarily high temperature within the lunar
257 mantle at this time, as conventional symmetric lunar thermal evolution models (Spohn et al., 2001; Ziethe et
258 al., 2009) predict much lower temperatures at 250 km depths at 2 Ga (~800–850 °C), under no circumstances
259 hot enough to cause mantle melting. Significant retention of internal heat in the Moon is predicted by thermal
260 models which include a combination of an insulating layer of megaregolith (i.e., Ziethe et al., 2009), a crustal
261 thickness of about 35 km and the highest concentration of heat producing elements in the mantle. The optimum
262 conditions for mantle melting in the relevant depth of ~250 km, however, do not persist until ~2 Ga, the time
263 of CE5 basalt generation (Ziethe et al., 2009).

264 More recent lunar thermal evolution models, that consider an asymmetric chemical composition of the
265 lunar mantle and crust with an elevated concentration of heat producing elements (KREEP+U, Th) in the
266 mantle right below the crust in the PKT region, lead to an asymmetric thermal evolution of the lunar mantle
267 and localized higher temperatures on the lunar nearside (Laneuville et al., 2013, 2018; Zhang et al., 2022). The
268 highest melt production at depths of 600 km depth starts at 4.5 Ga with a peak between 4–3.5 Ga ceasing at
269 0.2 Ga within the PKT region (Laneuville et al., 2013). Similarly, the model of Laneuville et al. (2018) predicts
270 volcanic activity on the lunar nearside until ~2 Ga, given enhanced heat production in the mantle beneath the
271 PKT. Generally, these models still tend to overestimate melt volume and underestimate the period of volcanic
272 activity on the Moon.

273 Given the experimentally determined minimum P (1.3 ± 0.1 GPa) and T conditions (1225 ± 25 °C) of
274 melting of the lunar mantle for CE5 basalt formation, we find that such an asymmetric thermal evolution of
275 the lunar interior is a prerequisite for prolonged volcanic activity on the lunar nearside (Laneuville et al., 2013,
276 2018; Zhang et al., 2022). This asymmetry can potentially be linked to the South Pole-Aitken basin (SPA)
277 impact (Zhang et al., 2022). This giant impact would have caused a thermochemical instability which pushed

278 the lunar mantle cumulates and the associated KREEP material towards the lunar nearside (Zhang et al., 2022)
279 and hence, caused or amplified a lunar magma ocean overturn (Ringwood and Kesson, 1976; Snyder et al.,
280 1992). The dynamic mixing and secondary melting due to concentration of heat-producing elements might
281 have formed a heterogeneous source necessary for the long-lasting period of mare volcanism (Laneuville et
282 al., 2013, 2018; Zhong et al., 2000). However, the impact dates back to 4.2–4 Ga and thus cannot be the direct
283 heat source for melting of Fe- and Ti-rich cumulates that must have formed the young CE5 basalts. The young
284 CE5 basalts testify that mantle melting happened at ~2 Ga. Yet, the absence of significant KREEP signatures
285 in the CE5 samples (e.g., Tian et al., 2021) demand for another explanation, then purely KREEP involvement
286 in the source. The distribution of this KREEP component in the lunar mantle, whatever its exact nature may
287 be, has to be very heterogeneous (Borg et al., 2009).

288 In conclusion, we state that the CE5 basalts could be primary melts of a heterogeneous late stage
289 mantle cumulate, but only if the thermal state of the lunar interior is highly asymmetric leading to a lunar
290 mantle with much less heat loss than in a symmetric lunar mantle. Note, however, that current asymmetric
291 thermodynamic evolution models of the Moon still fail to explain the long-lasting volcanic activity up until ~
292 1 Ga and more work is clearly needed on these matters (Laneuville et al. 2013, 2018).

293

294 Second, our fractional crystallization model defines a potential source composition of the CE5 basalts with
295 minimum melting conditions of >1500 °C and 2.8–2.5 GPa (Brown and Grove, 2015; Fig. 4), Corresponding
296 to a depth of some 500 km deep within the lunar mantle. If the CE5 basalts were products of fractional
297 crystallization, as proposed by e.g. Tian et al. (2021), the mantle at 500 km depth must have had a temperature
298 of at least 1530 °C around the formation age (~2.0 Ga) of the parental melts of the CE5 basalts (Fig. 4). Based
299 on the discussion of the thermal evolution (see above), it is highly unlikely that a lunar mantle retained such
300 high temperatures (> 1500 °C) until 2 Ga; Ziethe et al., 2009; Laneuville et al., 2013, 2018).

301 In summary, it is unlikely that the CE5 basalts formed by fractional crystallization from a primitive
302 melt similar to the A14 yellow glass composition, since temperatures of >1500 °C in the lunar mantle at the
303 time of CE5 basalt formation are beyond any thermodynamic assumption (Fig. 4b). Lunar mantle temperatures

304 of >1200 °C, as indicated by our MSP experiments, are much more likely in the lunar shallow mantle at, or
305 close to the time of eruption of the CE5 basalts at 2 Ga, and hence, we argue that these rocks are most likely
306 product of melting of late-stage, Fe- and Ti-rich cumulates (Fig. 4a).

307

308

309 **Acknowledgements**

310 Our thanks go to Maik Trogisch for his excellent sample preparation and to Beate Schmitte for her superb
311 help with the EPMA measurements. This work was funded by the Deutsche Forschungsgemeinschaft (DFG,
312 German Research Foundation)–Project ID 263649064–TRR 170. This is publication number xxx. MK
313 acknowledges a postdoctoral research fellowship from the Humboldt Foundation. We would also like to
314 thank Clive Neal and an anonymous reviewer for their constructive and very helpful reviews.

315

316

317

318

319

320

321

322

323

324

- 326 Armstrong, J.T., 1991. Quantitative elemental analysis of individual microparticles with electron beam
327 instruments, in: *Electron Probe Quantitation*. Springer, pp. 261–315.
- 328 Borg, L.E., Gaffney, A.M., Shearer, C.K., DePaolo, D.J., Hutcheon, I.D., Owens, T.L., Ramon, E., Brenneka,
329 G., 2009. Mechanisms for incompatible-element enrichment on the Moon deduced from the lunar
330 basaltic meteorite Northwest Africa 032. *Geochim. Cosmochim. Acta* 73, 3963–3980.
331 <https://doi.org/10.1016/j.gca.2009.03.039>
- 332 Borg, L.E., Shearer, C.K., Asmerom, Y., Papike, J.J., 2004. Prolonged KREEP magmatism on the Moon
333 indicated by the youngest dated lunar igneous rock. *Nature* 432, 209–211.
334 <https://doi.org/10.1038/nature03070>
- 335 Bose, K., Ganguly, J., 1995. Quartz-coesite transition revisited: reversed experimental determination at 500–
336 1200 C and retrieved thermochemical properties. *Am. Mineral.* 80, 231–238.
- 337 Boyd, F.R., England, J.L., 1960. Apparatus for phase-equilibrium measurements at pressures up to 50
338 kilobars and temperatures up to 1750°C. *J. Geophys. Res.* 1896-1977 65, 741–748. <https://doi.org/10.1029/JZ065i002p00741>.
- 340 Brown, S.M., Grove, T.L., 2015. Origin of the Apollo 14, 15, and 17 yellow ultramafic glasses by mixing of
341 deep cumulate remelts. *Geochim. Cosmochim. Acta* 171, 201–215. <https://doi.org/10.1016/j.gca.2015.09.001>
- 342 Che, X., Nemchin, A., Liu, D., Long, T., Wang, C., Norman, M.D., Joy, K.H., Tartese, R., Head, J., Jolliff, B.,
343 Snape, J.F., Neal, C.R., Whitehouse, M.J., Crow, C., Benedix, G., Jourdan, F., Yang, Z., Yang, C.,
344 Liu, J., Xie, S., Bao, Z., Fan, R., Li, D., Li, Z., Webb, S.G., 2021. Age and composition of young
345 basalts on the Moon, measured from samples returned by Chang'e-5. *Science* 374, 887–890.
346 <https://doi.org/10.1126/science.abl7957>
- 347 Chen, H.-K., Delano, J.W., Lindsley, D.H., 1982. Chemistry and phase relations of VLT volcanic glasses from
348 Apollo 14 and Apollo 17. *J. Geophys. Res. Solid Earth* 87, A171–A181.
349 <https://doi.org/10.1029/JB087iS01p0A171>
- 350 Day, J.M.D., Taylor, L.A., Floss, C., Patchen, A.D., Schnare, D.W., Pearson, D.G., 2006. Comparative
351 petrology, geochemistry, and petrogenesis of evolved, low-Ti lunar mare basalt meteorites from the
352 LaPaz Icefield, Antarctica. *Geochim. Cosmochim. Acta* 70, 1581–1600.
353 <https://doi.org/10.1016/j.gca.2005.11.015>
- 354 Delano, J.W., 1986. Pristine lunar glasses: criteria, data, and implications. *J. Geophys. Res. Solid Earth* 91,
355 201–213. <https://doi.org/10.1029/JB091iB04p0D201>
- 356 Delano, J.W., Lindsley, D.H., 1983. Mare glasses from Apollo 17: constraints on the Moon's bulk composition.
357 *J. Geophys. Res. Solid Earth* 88, B3–B16. <https://doi.org/10.1029/JB088iS01p000B3>
- 358 Elardo, S.M., Shearer, C.K., Fagan, A.L., Borg, L.E., Gaffney, A.M., Burger, P.V., Neal, C.R., Fernandes,
359 V.A., McCubbin, F.M., 2014. The origin of young mare basalts inferred from lunar meteorites
360 Northwest Africa 4734, 032, and LaPaz Icefield 02205. *Meteorit. Planet. Sci.* 49, 261–291.
361 <https://doi.org/10.1111/maps.12239>
- 362 Elardo, S.M., Shearer, C.K., Vander Kaaden, K.E., McCubbin, F.M., Bell, A.S., 2015. Petrogenesis of
363 primitive and evolved basalts in a cooling Moon: experimental constraints from the youngest known
364 lunar magmas. *Earth Planet. Sci. Lett.* 422, 126–137. <https://doi.org/10.1016/j.epsl.2015.04.014>
- 365 Elkins-Tanton, L.T., Burgess, S., Yin, Q.-Z., 2011. The lunar magma ocean: reconciling the solidification
366 process with lunar petrology and geochronology. *Earth Planet. Sci. Lett.* 304, 326–336.
367 <https://doi.org/10.1016/j.epsl.2011.02.004>
- 368 Green, D.H., Ringwood, A.E., Hibberson, W.O., Ware, N.G., 1975. Experimental petrology of Apollo 17 mare
369 basalts, in: *Lunar and Planetary Science Conference Proceedings*. pp. 871–893.
- 370 Green, D.H., Ringwood, A.E., Ware, N.G., Hibberson, W.O., Major, A., Kiss, E., 1971. Experimental
371 petrology and petrogenesis of Apollo 12 basalts, in: *Lunar and Planetary Science Conference*
372 *Proceedings*. p. 601.

373 Grove, T.L., Vaniman, D.T., 1978. Experimental petrology of very low TI/VLT/basalts, in: Mare Crisium: The
374 View from Luna 24. pp. 445–471.

375 Hiesinger, H., Head III, J.W., Wolf, U., Jaumann, R., Neukum, G., 2003. Ages and stratigraphy of mare basalts
376 in Oceanus Procellarum, Mare Nubium, Mare Cognitum, and Mare Insularum. *J. Geophys. Res.*
377 *Planets* 108. <https://doi.org/10.1029/2002JE001985>

378 Hu, S., He, Huicun, Ji, J., Lin, Y., Hui, H., Anand, M., Tartèse, R., Yan, Y., Hao, J., Li, R., Gu, L., Guo, Q.,
379 He, Huaiyu, Ouyang, Z., 2021. A dry lunar mantle reservoir for young mare basalts of Chang'E-5.
380 *Nature* 1–8. <https://doi.org/10.1038/s41586-021-04107-9>

381 Hughes, S.S., Delano, J.W., Schmitt, R.A., 1988. Apollo 15 yellow-brown volcanic glass: chemistry and
382 petrogenetic relations to green volcanic glass and olivine-normative mare basalts. *Geochim.*
383 *Cosmochim. Acta* 52, 2379–2391. [https://doi.org/10.1016/0016-7037\(88\)90295-5](https://doi.org/10.1016/0016-7037(88)90295-5)

384 Klemme, S., O'Neill, H.S., 1997. The reaction $MgCr_2O_4 + SiO_2 = Cr_2O_3 + MgSiO_3$ and the free energy
385 of formation of magnesiochromite ($MgCr_2O_4$). *Contrib. Mineral. Petrol.* 130, 59–65.

386 Krawczynski, M.J., Grove, T.L., 2012. Experimental investigation of the influence of oxygen fugacity on the
387 source depths for high titanium lunar ultramafic magmas. *Geochim. Cosmochim. Acta* 79, 1–19.
388 <https://doi.org/10.1016/j.gca.2011.10.043>

389 Laneuville, M., Wiczorek, M.A., Breuer, D., Tosi, N., 2013. Asymmetric thermal evolution of the Moon. *J.*
390 *Geophys. Res. Planets* 118, 1435–1452. <https://doi.org/10.1002/jgre.20103>

391 Laneuville, M., Taylor, J., & Wiczorek, M. A., 2018. Distribution of radioactive heat sources and
392 thermal history of the Moon. *J. Geophys. Res. Planets*, 123, 3144–3166.

393 Li, C., Hu, H., Yang M.-F., Pei, Z.-Y., Zhou, Q., Ren, X., Liu, B., Zeng, X., Zhang, G., Zhang, H., et al., 2022.
394 Characteristics of the lunar samples returned by the Chang'E-5 mission. *Natl. Sci. Rev.*, 9, nwab188.

395 Liu, X., Hao, J., Li, R., He, Y., Tian, H., Hu, S., Li, J., Gu, L., Yang, W., Lin, Y., 2022. Sulfur isotopic
396 fractionation of the youngest Chang'e-5 basalts: constraints on the magma degassing and geochemical
397 features of the mantle source. *Geophys. Res. Lett.* 49. <https://doi.org/10.1029/2022GL099922>

398 Longhi, J., 1992a. Origin of picritic green glass magmas by polybaric fractional fusion, in: *Lunar and Planetary*
399 *Science Conference Proceedings*. pp. 343–353.

400 Longhi, J., 1992b. Experimental petrology and petrogenesis of mare volcanics. *Geochim. Cosmochim. Acta*
401 56, 2235–2251. [https://doi.org/10.1016/0016-7037\(92\)90186-M](https://doi.org/10.1016/0016-7037(92)90186-M)

402 McCubbin, F.M., Steele, A., Hauri, E.H., Nekvasil, H., Yamashita, S., Hemley, R.J., 2010. Nominally hydrous
403 magmatism on the Moon. *Proc. Natl. Acad. Sci.* 107, 11223–11228.
404 <https://doi.org/10.1073/pnas.1006677107>

405 Papike, J.J., Vaniman, D.T., 1978. Luna 24 ferrobasalts and the mare basalt suite: comparative chemistry,
406 mineralogy and petrology. *Mare Crisium View Luna 24*.

407 Qian, Y., Xiao, L., Wang, Q., Head, J.W., Yang, R., Kang, Y., van der Bogert, C.H., Hiesinger, H., Lai, X.,
408 Wang, G., Pang, Y., Zhang, N., Yuan, Y., He, Q., Huang, J., Zhao, J., Wang, J., Zhao, S., 2021. China's
409 Chang'e-5 landing site: geology, stratigraphy, and provenance of materials. *Earth Planet. Sci. Lett.*
410 561, 116855. <https://doi.org/10.1016/j.epsl.2021.116855>

411 Qian, Y.Q., Xiao, L., Zhao, S.Y., Zhao, J.N., Huang, J., Flahaut, J., Martinot, M., Head, J.W., Hiesinger, H.,
412 Wang, G.X., 2018. Geology and scientific significance of the Rümker region in northern Oceanus
413 Procellarum: China's Chang'E-5 landing region. *J. Geophys. Res. Planets* 123, 1407–1430.
414 <https://doi.org/10.1029/2018JE005595>

415 Rhodes, J.M., Blanchard, D.P., Dungan, M.A., Brannon, J.C., Rodgers, K.V., 1977. Chemistry of Apollo 12
416 mare basalts-Magma types and fractionation processes, in: *Lunar and Planetary Science Conference*
417 *Proceedings*. pp. 1305–1338.

418 Ringwood, A.E., Kesson, S.E., 1976. A dynamic model for mare basalt petrogenesis. *Lunar Planet. Sci. Conf.*
419 *Proc.* 7.

420 Sato, M., Hickling, N.L., McLane, J.W., 1973. Oxygen fugacity values of Apollo 12, 14 and 15 lunar samples
421 and reduced state of lunar magmas. *Proc. Fourth Lunar Sci. Conf. Suppl.* 4 *Geochim. Cosmochim.*
422 *Acta* 1, 1061–1079.

423 Schmidt, M.W., Kraettli, G., 2022. Experimental crystallization of the lunar magma ocean, initial selenotherm
424 and density stratification, and implications for crust formation, overturn and the bulk silicate Moon
425 composition. *J. Geophys. Res. Planets*, 127, e2022JE007187.

426 Shearer, C.K., Hess, P.C., Wieczorek, M.A., Pritchard, M.E., Parmentier, E.M., Borg, L.E., Longhi, J., Elkins-
427 Tanton, L.T., Neal, C.R., Antonenko, I., Canup, R.M., Halliday, A.N., Grove, T.L., Hager, B.H., Lee,
428 D.-C., Wiechert, U., 2006. Thermal and magmatic evolution of the Moon. *Rev. Mineral. Geochem.*
429 60, 365–518. <https://doi.org/10.2138/rmg.2006.60.4>

430 Singletary, S., Grove, T., 2008. Origin of lunar high-titanium ultramafic glasses: a hybridized source? *Earth*
431 *Planet. Sci. Lett.* 268, 182–189. <https://doi.org/10.1016/j.epsl.2008.01.019>

432 Snyder, G.A., Taylor, L.A., Neal, C.R., 1992. A chemical model for generating the sources of mare basalts:
433 combined equilibrium and fractional crystallization of the lunar magmasphere. *Geochim. Cosmochim.*
434 *Acta* 56, 3809–3823. [https://doi.org/10.1016/0016-7037\(92\)90172-F](https://doi.org/10.1016/0016-7037(92)90172-F)

435 Spera, F.J., 1992. Lunar magma transport phenomena. *Geochim. Cosmochim. Acta* 56, 2253–2265.

436 Spohn, T., Konrad, W., Breuer, D., Ziethe, R., 2001. The longevity of lunar volcanism: implications of thermal
437 Evolution Calculations with 2D and 3D Mantle Convection Models. *Icarus* 149, 54–65.
438 <https://doi.org/10.1006/icar.2000.6514>

439 Tian, H.-C., Wang, H., Chen, Y., Yang, W., Zhou, Q., Zhang, C., Lin, H.-L., Huang, C., Wu, S.-T., Jia, L.-H.,
440 Xu, L., Zhang, D., Li, X.-G., Chang, R., Yang, Y.-H., Xie, L.-W., Zhang, D.-P., Zhang, G.-L., Yang,
441 S.-H., Wu, F.-Y., 2021. Non-KREEP origin for Chang’E-5 basalts in the Procellarum KREEP Terrane.
442 *Nature* 1–9. <https://doi.org/10.1038/s41586-021-04119-5>

443 van Orman, J.A., Grove, T.L., 2000. Origin of lunar high-titanium ultramafic glasses: constraints from phase
444 relations and dissolution kinetics of clinopyroxene-ilmenite cumulates. *Meteorit. Planet. Sci.* 35, 783–
445 794. <https://doi.org/10.1111/j.1945-5100.2000.tb01462.x>

446 Wadhwa M., 2008. Redox conditions on small bodies, the Moon and Mars. *Rev. Mineral. Geochem.* 68, 493–
447 510.

448 Wagner, T.P., Grove, T.L., 1997. Experimental constraints on the origin of lunar high-Ti ultramafic glasses.
449 *Geochim. Cosmochim. Acta* 61, 1315–1327. [https://doi.org/10.1016/S0016-7037\(96\)00387-0](https://doi.org/10.1016/S0016-7037(96)00387-0)

450 Wieczorek, M.A., Neumann, G.A., Nimmo, F., Kiefer, W.S., Taylor, G.J., Melosh, H.J., Phillips, R.J., Solomo,
451 S.C., Andrews-Hanna, J.C., Asmar, S.W., Konopliv, A.S., Lemoine, F.G., Smith, D.E., Watkins,
452 M.M., Williams, J.G., Zuber, M.T., 2013. The crust of the Moon as Seen by GRAIL. *Sci. Rep.* 339,
453 671–675.

454 Zhang, N., Ding, M., Zhu, M.-H., Li, Huacheng, Li, Haoyuan, Yue, Z., 2022. Lunar compositional asymmetry
455 explained by mantle overturn following the South Pole–Aitken impact. *Nat. Geosci.* 15, 37–41.
456 <https://doi.org/10.1038/s41561-021-00872-4>

457 Zhong, S., Parmentier, E.M., Zuber, M.T., 2000. A dynamic origin for the global asymmetry of lunar mare
458 basalts. *Earth Planet. Sci. Lett.* 177, 131–140. [https://doi.org/10.1016/S0012-821X\(00\)00041-8](https://doi.org/10.1016/S0012-821X(00)00041-8)

459 Ziethe, R., Seiferlin, K., Hiesinger, H., 2009. Duration and extent of lunar volcanism: comparison of 3D
460 convection models to mare basalt ages. *Planet. Space Sci.* 57, 784–796.

461 Zong, K., Wang, Z., Li, J., He, Q., Li, Y., Becker, H., Zhang, W., Hu, Z., He, T., Cao, K., She, Z., Wu, X.,
462 Xiao, L., Liu, Y., 2022. Bulk compositions of the Chang’E-5 lunar soil: insights into chemical
463 homogeneity, exotic addition, and origin of landing site basalts, *Geochim. Cosmochim. Acta* 335,
464 284–296,
465